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Key indicators

Single-crystal X-ray study T = 143 K Mean σ (C–C) = 0.001 Å R factor = 0.026 wR factor = 0.069 Data-to-parameter ratio = 25.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N,N-Bis(trifluoromethylsulfonyl)aniline

The title molecule, $C_8H_5F_6NO_4S_2$, has crystallographic twofold symmetry. The crystal packing consists of three intermolecular C-H···O contacts, resulting in a two-dimensional network. In the third direction, only a very weak C-H···F interaction, with an H···F distance of 2.83 Å, is found.

Comment

The crystal structure determination of the title compound, (I), was undertaken to study the influence of the F atoms on the crystal packing.



The molecule (Fig. 1) has C2 symmetry: atoms N, C1, C4 and H4 lie on a crystallographic twofold axis. This twofold symmetry results in an exact planarity of the N atom. The angle between the plane defined by the N atom and the plane of the phenyl group is 80.08 (3)°. The molecule shows no short intramolecular contacts. A very similar almost C2-symmetrical molecular conformation, with an angle of 85.2° between the planes defined by the N atom and the phenyl group, has been reported for N,N-bis(methylsulfonyl)aniline (Jones *et al.*, 1995).

The crystal packing (Fig. 2 and Table 1) shows three intermolecular C-H···O interactions. The C4-H4 bond donates a bifurcated weak hydrogen bond in the **c** direction to O atoms of two symmetry-related molecules. The molecules are connected in the **b** direction by a C2-H2···O2 interaction. No short intermolecular contacts are found which connect the molecules in the **a** direction. The shortest interaction found in this direction is a C3-H3···F3(1/4-x, 3/4+y, z-1/4) contact, with an H···F distance of 2.83 Å. This contact is very long and thus should be rather weak.

Experimental

© 2002 International Union of Crystallography Printed in Great Britain – all rights reserved A commercial sample of (I) (Aldrich Chemical Co.) was recrystallized from ether at 293 K. Received 19 April 2002 Accepted 26 April 2002

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Crystal data

C₈H₅F₆NO₄S₂ $M_r = 357.25$ Orthorhombic, Fdd2 a = 24.871 (4) Å b = 5.6208 (6) Å c = 18.2216 (19) Å V = 2547.3 (6) Å² Z = 8 $D_x = 1.863 \text{ Mg m}^{-3}$

Data collection

Siemens SMART CCD diffractometer ω scans Absorption correction: numerical (SHELXTL; Sheldrick, 1996) $T_{\min} = 0.823, \ T_{\max} = 0.946$ 13534 measured reflections 2529 independent reflections 2334 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.026$ $wR(F^2) = 0.069$ S=1.202529 reflections 98 parameters H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.04P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$

Mo $K\alpha$ radiation Cell parameters from 234 reflections $\theta = 3 - 36^{\circ}$ $\mu = 0.51 \text{ mm}^{-1}$ T = 143 (2) KRod, colorless $0.50 \times 0.18 \times 0.12 \text{ mm}$

 $R_{\rm int} = 0.029$ $\theta_{\rm max} = 35.0^{\circ}$ $h = -33 \rightarrow 39$ $k=-8\rightarrow 9$ $l = -28 \rightarrow 26$ 205 standard reflections frequency: 600 min intensity decay: none

 $(\Delta/\sigma)_{\rm max} = 0.001$ $\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$ $\Delta \rho_{\rm min} = -0.24 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL97 Extinction coefficient: 0.00060 (14) Absolute structure: Flack (1983); 1139 Friedel pairs Flack parameter = 0.00(5)



Figure 2 The crystal packing of (I), viewed down b.

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C2-H2\cdots O2^{i}$	0.95	2.53	3.458 (2)	167
$C4-H4\cdots O1^{ii}$	0.95	2.51	3.267 (2)	137

Symmetry codes: (i) x, 1 + y, z; (ii) $x, \frac{1}{2} + y, z - \frac{1}{2}$.

The H atoms were taken from a difference Fourier synthesis. They were refined with fixed individual displacement parameters $[U_{iso}(H)]$ = $1.2U_{eq}(C)$], using a riding model with fixed distances (H-C = 0.95 Å).

Data collection: SMART (Siemens, 1995); cell refinement: SMART; data reduction: SAINT (Siemens, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in SHELXTL (Sheldrick, 1996); software used to prepare material for publication: SHELXL97.

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